# CYCLIZATION REACTIONS OF QUATERNARY SALTS OF AZAAROMATICS

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*Cyclization reactions of quaternary salts of azaaromatics including inter- and intramolecular condensation as well as radical processes are described.* 

#### I. INTRODUCTION

Numerous cyclization reactions of azaaromatic quaternary salts axe known [1-10]; they often provide a convenient approach to fused heterocycles. In the continuation of our research concerning azaaromatics [11-14] and their quaternary salts [15-21] we present here chosen examples of cyclization reactions of these compounds. As cycloaddition reactions of azaaromatics cover a large amount of works [22-25], this topic is not included here.

#### H. INTERMOLECULAR CYCLOCONDENSATION REACTIONS

The  $[4 + 1]$  cyclocondensation reaction of 2-alkyl- and 2-aminoazaaromatic salts with suitable phosphorus reagents is a facile synthetic route to a series of armelated azaphospholes with a bridgehead nitrogen atom [1, 26]. The cyclocondensation of 1,2-disubstituted pyridinium bromides (1) with phosphorus trichloride in the presence of triethylamine leads to 2 phosphaindolizines (2). It was observed that the cyclization occurs when both N-methylene and 2-alkyl groups are activated by electron-withdrawing substituents [27, 28].



In order to obtain 1-unsubstituted product (3), an analogous reaction with the use of salt 4 as starting material has been performed [28]. The synthesis of the 1-unsubstituted 2-phosphaindolizine 3 is often accompanied by the formation of 1 dichlorophosphino derivative (5), which is a secondary product arising from the electrophilic substitution of 3 by phosphorus trichloride [29]. When the reaction is carried out in acetonitrile at  $0^{\circ}$ C, the formation of 5 may be avoided [28].



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**[28].**  From 5,6,7,8-tetrahydro-l-phenacylquinolinium bromide (6) the tricyclic phosphaindolizine derivative 7 is formed



In order to synthesize 3-ethoxycarbonyl-l,4,2-diazaphospholopyridines (8), 2-aminopyridinium bromides 9 were treated with phosphorus trichloride and triethylamine; depending on the  $R<sup>1</sup>$  substituent, the reaction affords diazaphospholopyridines 8 or stops on the formation of their intermediates, aminodichlorophosphines 10 [30, 31].



c  $R^1$  = COOEt,  $R^2$  =  $R^3$  = H,  $R^4$  = Me,  $X$  = Br; d  $R^1$  = COOEt,  $R^2$  = Me,  $R^3$  =  $R^3$  = H,  $X$  = Br; e  $R^2$  = Ph;  $R^2 = R^3 = R^4 = H$ ,  $X = Cl$ ; f  $R^1 = 4$ -MeC<sub>6</sub>H<sub>4</sub>,  $R^2 = R^3 = H$ ,  $X = Br$ 

For ring closure, the presence of the electron-withdrawing  $R<sup>t</sup>$  substituent, here COOEt, is necessary; then 8a-d are obtained; however, in the case of  $R^1 = Ph$  or 4-MeC<sub>6</sub>H<sub>4</sub>, the reaction leads only to 10e,f, respectively [30].

Reactions of this type, proceeding on quaternary salts of heterocycles other than pyridine, should be mentioned here also, for example:



R = CN, COPh, COOEt

Bicyclic systems of the type 11 with bridgehead quaternary nitrogen atom can be easily obtained by the Westphal condensation, i.e., the reaction of 1,2-disubstituted pyridinium salts with 1,2-diketones in the presence of a base [2, 35-37].



The Westphal condensation of pyridinium salts 12a-e with 1,2-acenaphthenequinone proceeds via zwitterions 13a-e, undergoing an intramolecular Claisen process to yield cyclazinone derivatives 14a-e isolated as their hydrobromides 15a-e [36].



 $12-15$  a  $R^1 = R^2 = R^3 = R^4 = R^5 = H$ ,  $X = Br$ ; b  $R^1 = R^2 = H$ ,  $R^3 = Me$ ,  $R^4 = R^5 = H$ ,  $X = Br$ ;  $c R^{1} = R^{2} = H, R^{3} = Ph, R^{4} = R^{5} = H, X = Br; d R^{1} = R^{2} = R^{3} = H, R^{4} = R^{5} = -(CH_{2})_{3} = K = Br;$ e  $R^1 - R^2 =$  --(CH<sub>2</sub>)<sub>3</sub>--,  $R^3 - H$ ,  $R^4 - R^5 =$  --(CH<sub>2</sub>)<sub>3</sub>--,  $X - BF_4$ ; f  $R^1 - OCOMe$ ,  $R^2 - R^3 - R^4 - R^5 - H$ ,  $X - Br$ 

The reaction of 12f with 1,2-acenaphthenequinone affords quinolizinium 1-olate (16) converted by 40% hydrobromic acid into 17. The betaine 18 derived from 17 was submitted to cycloaddition reaction with DMAD resulting in [2.3.3.]cyclazin-6-one 19 [36].



Further examples of Westphal condensation are given in [38].

We should note here also the cyclocondensation of quaternary pyridinium salts 20 with 1,8-diaminonaphthalene resulting in (perimidinylmethyl)-pyridinium salts 21, converted by treatment with phenyl isothiocyanate into 22 [39].



## **III. INTRAMOLECULAR CYCLOCONDENSATION REACTIONS**

The intramolecular cyclocondensation is another type of cyclization reaction, often used in the syntheses of polyheterocycles.

The reaction of  $\alpha$ -picoline or 2-aminopyridine with 1-adamantyl-2-bromomethylketone leads to quaternary salts 23 undergoing the intramolecular cyclocondensation to afford 24a or 24b, respectively [40].



Pyridinium salts 25 were cyclized into indolizines 26 by treatment with NaHCO<sub>3</sub>. Compounds 26 can be converted by sodium hydroxide into indoles 27 via ring opening and recyclization [41, 42].



It was observed that quaternary salts  $28$  at pH ca. 7 (treatment with KHCO<sub>3</sub>) give ylides  $29$ , which are stable on storage in air. Systems 28 or 29 at pH 8-10 (treatment with  $K_2CO_3$ ) yield 1,3-dioxanoindolizines 30. The mechanism of this process involves the tautomerization of 29 resulting in zwitterions 31, which undergo a subsequent aldol condensation [43].





 $Ar = Ph$ , 4-BrC<sub>6</sub>H<sub>4</sub>;  $R^1 = H$ , Me;  $R^2 = OH$ , OAc

The reaction of 1-ethoxycarbonylmethyl-2-methylpyridinium bromide with sodium hydroxide in ethanol leads to the formation of enamino ester 32, rearranging into pyrroloquinolone 33 [44].



We should mention here also the reaction of derivatives of 1-ethoxycarbonylmethylpyridinium bromide with sodium ethoxide resulting in indolizinones 34, which when treated *in situ* with methyl dithioacetate or methyl dithiobenzoate afford smoothly mercaptomethyleneindolizinones 35 [45].



Pyranoindolizinones (36) are formed as minor products in the process shown below; the mechanism of this reaction is given [46].



 $36.9 P<sup>1</sup>$  – Me,  $R<sup>2</sup>$  – H; b R<sup>1</sup> – R<sup>2</sup> – Me; c R<sup>2</sup> – Me, R<sup>2</sup> – Et; d R<sup>2</sup> – Et, R<sup>2</sup> – H<sub>2</sub> **e R I - Et, R 2 - Me, f R I - R 2 - Et** 

The following cyclization has been performed [47].



The formation of the tricyclic ethanoadenosine 37 was observed in the Dimroth rearrangement of 1-(2 aminoethyl)adenosine 38 leading to  $N^6$ -(2-aminoethyl)adenosine 39 [48, 49]. The proposed mechanism of this reaction is given below.



 $R = D$ -Ribose;  $N^*$  = labeled <sup>15</sup>N

Pyridinium salts 40 easily undergo ring closure by ammonium acetate to give oxopyridopyrazinium salts 41. These compounds were converted into stable zwitterionic systems 42 [50].



40-42 a R<sup>1</sup> = Et, R<sup>2</sup> = t=Bu; b R<sup>1</sup> = Et, R<sup>2</sup> = C<sub>6</sub>H<sub>5</sub>; c R<sup>1</sup> = H, R<sup>2</sup> = 4-MeC<sub>6</sub>H<sub>4</sub>; d  $R^1$  = Et,  $R^2$  = 4-BrC<sub>6</sub>H<sub>4</sub>; e  $R^1$  = H,  $R^2$  = 4-NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>

In order to obtain benzo $[a]$ quinolizinium salts 43, 2-bromopicolines 44 were treated with thiophenol in the presence of triethylamine. The formed 2-(phenylthio)picolines 45 reacted with bromoacetaldehyde oxime in sulfolane to give the quaternary salts 46. The cyclization of 46 with 48% HBr afforded the thiazepinium salts 47, which by treatment with  $30\%$  H<sub>2</sub>O<sub>2</sub> in acetic acid underwent oxidative sulfur extrusion resulting in 43 [51].



[52]. Pyridiniothiophenes 48 submitted to dehydrobromination with DBU afford dihydropyridothienothiazines 49 and/or 50



The reaction of pyridines 51 with chloroacetylanthranilonitriles 52 leads to pyridinioquinolinone chlorides 53; the intermediate pyridinium salts 54 are isolable; however, their isolation is not necessary [53].



The deprotonation of  $53a$ ,c with  $\sim$  2 N NaOH yields pyridoimidazoquinolones 55; the reaction proceeds via pyridinioquinolinolates 56, immediately undergoing an intramolecular addition of amino group to the pyridinium ring, followed by the dehydrogenation with air oxygen [53, 54]. Compound 55 may also be obtained from 53 by cyclooxidation with bromine [53].



The following reactions involving nucleophilic addition of the amino group to the pyridinium ring and subsequent dehydration lead to stable products 57 [3]:



57 a R<sup>1</sup> – H, R<sup>2</sup> – Me, R<sup>3</sup> – NH<sub>2</sub>; b R<sup>1</sup> – NO<sub>2</sub>, R<sup>2</sup> – H, R<sup>2</sup> – NH<sub>2</sub>; c R<sup>2</sup> – R<sup>2</sup> – H, R<sup>2</sup> – morpholinyl

#### IV. OTHER CYCLIZATION REACTIONS

The reactions of nitropyridinium salts with ketimines or with a mixture of ketones and amines resulting in polyalkylindoles also belong to cyclization processes. It is a convenient approach to these heterocycles, difficult to obtain by other routes. The process occurs under mild conditions, and the products are easily isolated [55]..

The reaction proceeds via the scission of the pyridinium cation at the  $C_{(3)}-C_{(4)}$  bond; the  $C_{(2)}-C_{(3)}$  moiety of the pyridinium cation is involved in the construction of the pyrrole ring, and the  $C_{(4)}-C_{(5)}-C_{(6)}$  moiety forms the benzene ring [55-**60].** 

For example,



 $Y = H$ , Me; Z = Me; R<sup>2</sup> = H, Me, Et, i-Pr; R<sup>2</sup> = H, Me, Et, Pr, i-Pr, t-Bu, Ph; R  $-$  – H, Me, i-Pr; R  $-$  – H, Me, Et, i-Pr

It was established that the quaternary salts bearing bulky substituents such as  $i$ -Pr or  $t$ -Bu in positions 2, 4, or 6 also undergo this process; the bulkiness of substituents does not influence significantly the reaction rate [60]. The most convenient are bipolar aprotic solvents such as DMF or DMSO; in nonpolar or protic solvents the reaction is slow [61].

The reaction of pyridinium bromides 58 with pyridylketene dithioacetal 59 [62-65] yielding indolizinoquinolizines 60 and 61 is another example of cyclization of azaaromatic quaternary salts [66, 67]. When the reaction of 58a and 59 was carried

out in refluxing ethanol, only the ylide 62 was formed. The mechanism of the above processes involving double cyclization was proposed.



60, 61 a  $R^1 - R^2 - H$ ; b  $R^1 - R^2 - Me$ ; c  $R^1 - H$ ,  $R^2 - Me$ 

The reaction of isoquinolinium bromide 63 with 59 in the presence of  $K_2CO_3$  in refluxing ethanol affords benzindolizinoquinolizines 64 and 65 [66].



Another approach to indolizinoquinolizines 66 involves the treatment of pyridiniumketene dithioacetal 67 with pyridylacetonitrile 68 in the presence of triethylamine in EtOH/CHCl<sub>3</sub>. When the reaction of 67 and 68 was carried out in the presence of  $K_2CO_3$  in MeCN, unstable pyridinium oxoindolizine 69 was formed, which by reduction with NaBH<sub>4</sub> yielded tetrahydropyridylquinolizine 70 [66].



The nucleophilic addition of stabilized carbon nucleophiles to N-alkylpyridinium salts is used in the synthesis of bridged indole alkaloids [4, 68, 69]. The first total synthesis of the alkaloid vinoxine, using this procedure, is shown below [70-72].



The reaction of the pyridinium salt 71 with enolate derived from acetylindole 72 affords 1,4-dihydropyridine 73. The treatment of 73 with HCl in benzene solution gives tetracyclic product 74a, and with [Me<sub>2</sub>SSMe]BF<sub>4</sub> or PhSeBr compounds 74b or 74c, respectively, are formed [73, 74]. Further treatment of 72c was useful in the synthesis of methylervitsine [73].



Similar reactions of pyridinium salts with other nucleophiles have been studied [75-77].

## **V. CYCLIZATION REACTIONS LEADING TO OLATE INNER SALTS**

Reactions of pyridinium salts leading to olate inner salts also deserve attention; some examples will be described here. The reaction of N-carbamoylmethylpyridinium chloride 75 with unsaturated nitriles 76a-f leads to olate inner salts 77a-f; they can also be obtained by condensation of N-ethoxycarbonylmethylpyridinium bromide 78 with aromatic aldehydes 79a-f and cyanoacetamide. Reactions are performed in the presence of triethylamine; pyridinium ylides which are intermediates in these processes undergo Michael addition [78].



It was established that the reaction of 78 and 76d in the presence of  $Et<sub>1</sub>N$  leads to the Michael adduct 80, i.e., to the pyridinium 1,4-ylide, and not to the ylide 81. The treatment of 80 with ammonia affords olate 77d via the intermediate 82.



The results of conformational analysis have shown that Michael adduct 80 and olates 77 are formed in a stereoselective way [78].

The pyridinium olate 83 was obtained from pyridinium ylide 84 by two routes  $(a)$  and  $(b)$  given below; route  $(b)$ involving a three-component condensation proved to be a more convenient one [79].



Two following reactions of pyridinium salts 85a,b leading regio- and stereoselectively to inner salts 86 have been performed [80, 81]:



85 a R = COOMe,  $X = Br$ ; b R = CONH<sub>2</sub>,  $X = Cl$ ; c COMe,  $X = Br$ ; d COPh,  $X = Br$ 

When 85c,d were treated with 87, the inner salts 88 were obtained [80].





Styrylpyridinium salts where the double bond is activated by the presence of pyridinium cation undergo nucleophilic addition to amines, thioles, and CH-acids; they also have the properties of highly stereoselective dienophiles. Reactions of substituted 1-styrylpyridinium salts with cyanothioacetamide 89 were studied.

When 1-styrylpyridinium bromides 90 react with 89 in ethanol in the presence of triethylamine, (method  $a$ ), thiolate inner salts 91 are formed via Michael addition. The treatment of the ylide 92 (obtained from 90c and malononitrile) with hydrogen sulfide in the presence of  $Et<sub>1</sub>N$  results also in 91c (method b) [82].

It was found that 91 can be obtained in a more convenient procedure, without the prior isolation of 1-styrylpyridinium salts, simply by condensing pyridinium salts 93a-c, and aldehydes 94 and 89 (method c); the reaction proceeds in a highly stereoselective manner [82].



90--94 a Ar" = 4-FC6H4, Ar" = Ph; b Ar" = 4-BrC6H4, Ar" = Ph; c Ar" = 3-C5H4N, Ar" = Ph; d ha" I - 4-FC6H4; Ar 2 - 4-BrC6H4

## VI. RADICAL CYCLIZATION REACTIONS

Intramolecular radical-induced substitution of pyridinium salts 95 resulting in bicyclic compounds 96 proceeds with the use of tributylstannyl radicals, in MeCN/THF [83, 84].



These reactions are of interest for alkaloid syntheses. It is worth noting that in the isolation of products chromatography is not necessary, due to the use of the lipophilic tributyltin hydride as radical source.

The intramolecular photocyclization reactions of the quaternary nvridinium ions  $97-99$  have been investigated [5].



The  $\pi$ -electron density, atom self-polarizability values, as well as free valences on each atom of 97, 98, and 99 have been calculated using a simple Hückel method. The photocyclization occurs via radical intermediates by breaking a C-Cl bond.

The C--CI bond breaking is facilitated by  $\pi$ -complex formation between chlorine atom and the  $\pi$ -electrons of the neighboring ring in the triplet state. When the  $\pi$ -complex is strong, the C--Cl bond breaking is easier, and the photocyclization rate increases [5].

The following photocyclization reactions have been performed:



Benzo[a]quinolizinium perchlorates (100) have been obtained by photocyclization of the appropriate N-styrylpyridinium bromides (101) in ethanol in the presence of iodine ( $\lambda > 280$  nm) [51].



100, 101 a  $R^1$  – Me,  $R^2 - R^3$  – H; b  $R^1 - R^3$  – H,  $R^2$  – Me; c  $R^1 - R^2$  – H,  $R^3$  – Me

## VII. CONCLUSION

Cyclization reactions of quaternary salts of azaaromatics are a convenient approach to heteropolycycles containing one or more nitrogen atoms. In view of the interesting properties of these compounds, for example, their use as biologically active species, dyes, or redox systems, promising in the construction of electronic devices, their reactions involving cyclization processes deserve attention.

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